

Letters to the Editor

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THE ULTRAVIOLET ABSORPTION SPECTRUM OF PARA FLUOROCHLOROBENZENE.

S. L. N. G. KRISHNAMACHARI

PHYSICS DEPARTMENT, ANDHRA UNIVERSITY, WALT AIR

(Received for publication, April 23, 1956)

In continuation of the work on the ortho- and meta-fluorochlorobenzenes (author, 1955, 1956), the ultraviolet absorption spectrum of para-fluorochlorobenzene in the vapour was studied. There is no previous report on the ultraviolet absorption of this compound. The spectrum in the present investigations was photographed with path lengths of 25, 50 and 75 cms. and at different temperatures ranging from -80°C to about $+100^{\circ}\text{C}$.

Two regions of absorption were observed : (1) a continuous one below 2150Å and (2) a discrete one in the region 2940-2375Å. These two regions merge together at higher vapour pressures. About 275 bands were measured in this discrete region. This system could be explained as due to the allowed electronic transition A_1-B_1 . In accordance with this, a strong 0,0 band and overtones and combinations of many totally symmetrical vibrations were observed. The strong band at 36276 cm^{-1} was chosen as the 0,0 band. Most of the bands were interpreted in terms of 5 upper state and 5 ground state frequencies. These, together with other data, are presented in the following table. On the red side of each of the strong bands, satellite bands were observed with frequency separations of 33 and 90 cm^{-1} , the former being more pronounced. The absorption spectrum

obtained with a path length of 50 cm and at -10°C is reproduced in the adjoining figure 1.

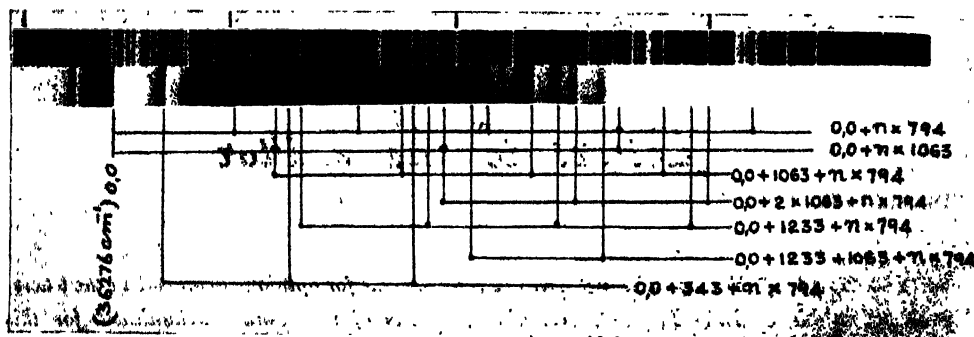


Fig. 1. Ultraviolet absorption spectrum of *p*-fluorochlorobenzene.

A detailed discussion of the analysis will be published shortly.

Ground and excited state frequencies of *p*-F.C₆H₄.Cl.

Raman data (Kohlrausch, 1947)			U.V. absorption data		Tentative assignment
$\lambda \nu$	Int	* ρ	Ground state	Excited state	
379	st	.42	370	343 (m)	one of the ϵ_g^+ com- ponents of C ₆ H ₆
634	ms	.64	637	563 (w)	C-Cl stretch
817	vs	.19	815	794 (st)	C-C vibration
1094	st	.22	1090	1063 (ms)	C-C vibration
1227	mw	.38	1239	1233 (m)	C-F stretch

*Determined in the present investigations.

The author is grateful to Dr. G. C. Finger of the Illinois State Geological Survey for the gift of the sample and to the Govt. of India for the award of a senior research scholarship. The author is deeply indebted to Prof. K. R. Rao for his valuable guidance.

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 Kohlrausch. *et al.*, 1947. *Mh. Fur. Chem.*, **76**, 200-14.